

## Design Features

- Austenitic structure.
- Very high resistance to pitting and crevice corrosion.
- Good resistance to uniform corrosion in environments containing halides.
- Very good resistance to stress corrosion cracking.
- Good formability and weldability.

## Applications

- Cooling water piping, heat exchangers.
- Pulp bleach plant components.
- FDG, Flue Gas Desulphurisation.
- Tall oil distillation.

Further information on the use of 254 SMO is given in reference lists, Inf. No. 9454, 9582, 10399GB.

## General Characteristics

The high levels of molybdenum in particular but also of chromium and nitrogen endow 254 SMO with extremely good resistance to pitting and crevice corrosion. The addition of copper provides improved resistance in certain acids. Furthermore, due to its relatively high nickel content in combination with the high levels of chromium and molybdenum 254 SMO possesses good resistance to stress corrosion cracking.

Even highly alloyed stainless steels have a very limited applicability in environments which contain halide ions, e.g.

chloride, bromide and fluoride solutions. For this reason it has often been necessary to use expensive materials such as nickel based alloys or titanium for these environments. 254 SMO was developed with a view to offering a less expensive alternative.

Numerous field tests and extensive application experience show that 254 SMO has a high resistance to crevice corrosion in seawater at ambient and slightly elevated temperatures. Very few stainless steels possess this property.

254 SMO resists attack in acidic, oxidising halide solutions that occur e.g. in pulp bleach plants and its resistance is such that the steel can often be ranked equal with the most resistant nickel based alloys and titanium.

Due to its high nitrogen content 254 SMO has higher mechanical strength than most other austenitic stainless steels. Like these 254 SMO is characterised by high ductility and impact strength as well as good weldability. The high molybdenum content of 254 SMO can lead to an increased oxidation rate at annealing and as a consequence a rougher surface finish after pickling than standard stainless steels.

Super alloy welding consumables, designated Avesta P12 and P16 respectively, are used for welding. These filler metals will give a weld with a pitting resistance comparable to the base metal.

Abrasive contact with copper/brass items should be avoided in the manufacturing process. If these substances are present in metallic form on the surface, they can cause surface cracks in connection with welding, hot forming and heat treatment.

## Chemical Composition

The chemical composition of specific steel grades may vary slightly between different national product standards. The required standard will be fully met as specified on the order.

Table 1

International steel No.		Avesta Sheffield steel name	Typical composition, %						National steel designations, superseded by EN			
EN	ASTM		C	N	Cr	Ni	Mo	Others	BS	DIN	NF	SS
1.4462	S31803	2205	0.02	0.17	22	5.7	3.1	–	318S13	1.4462	Z3 CND 22-05 Az	2377
1.4410	S32750	SAF 2507™	0.02	0.27	25	7	4	–	–	–	Z3 CND 25-06 Az	2328
1.4436	316	17-12-2.5	0.04	0.06	17	11	2.7	–	316S33	1.4436	Z7 CND 18-12-03	2343
1.4439	S31726	17-14-4LN	0.02	0.14	17.3	12.7	4.2	–	–	1.4439	Z3 CND 18-14-05 Az	–
1.4539	N08904	904L	0.01	0.06	20	25	4.5	1.5Cu	904S13	1.4539	Z2 NCDU 25-20	2562
<b>1.4547</b>	<b>S31254</b>	<b>254 SMO®</b>	<b>0.01</b>	<b>0.20</b>	<b>20</b>	<b>18</b>	<b>6.1</b>	<b>Cu</b>	–	–	–	<b>2378</b>
Ni-alloys*			Cmax									
–	N06007	Alloy G	0.05	–	22	44	6.5	Cu	–	2.4618	–	–
–	N06625	Alloy 625	0.10	–	21	60	9	Nb	–	2.4856	–	–
–	N10276	Alloy C-276	0.02	–	15	60	16	W	–	2.4819	–	–

\*These alloys are not produced within the Avesta Sheffield Group.

## Structure

254 SMO is annealed at 1150–1200°C to obtain an austenitic structure. In certain cases there may be traces of intermetallic phases (chi and sigma phase) in the centre of the material. Normally, however, these have no influence on impact strength or corrosion resistance. When exposed in the temperature range 600–1000°C these phases can precipitate at the grain boundaries. If the recommendations given for hot forming, welding and heat treatment are followed, there will be no precipitates affecting the corrosion resistance.

### TEMPERATURE CHARACTERISTICS

Table 2

	Temperature °C
<b>Solidification range</b>	<b>1400–1325</b>
<b>Scaling temperature in air</b>	<b>1000</b>
<b>Hot forming</b>	<b>1200–1000</b>
<b>Quench annealing</b>	<b>1150–1200 water*</b>
<b>Pressure vessel application</b>	<b>(–60)–400</b>

\*1120–1150 air/water for material in thicknesses below 2 mm.

## Mechanical Properties

Mechanical properties are given in Tables 3 and 4. Hot rolled plate in thicknesses 10–30 mm, transverse to the rolling direction.

Table 3

Minimum values			20°C
<b>Yield strength</b>	<b>R<sub>p0.2</sub></b>	<b>N/mm<sup>2</sup></b>	<b>300</b>
	<b>R<sub>p1.0</sub></b>	<b>N/mm<sup>2</sup></b>	<b>340</b>
<b>Tensile strength</b>	<b>R<sub>m</sub></b>	<b>N/mm<sup>2</sup></b>	<b>650</b>
<b>Elongation</b>	<b>A<sub>5</sub></b>	<b>%</b>	<b>35</b>
<b>Hardness</b>	<b>HB</b>	<b>max.</b>	<b>210</b>
<b>Impact value</b>	<b>KCV</b>	<b>J/cm<sup>2</sup></b>	<b>120</b>

### TENSILE PROPERTIES

#### AT ELEVATED TEMPERATURES

Table 4

Minimum values		50	100	200	300	400°C
<b>R<sub>p0.2</sub></b>	<b>N/mm<sup>2</sup></b>	<b>270</b>	<b>235</b>	<b>195</b>	<b>175</b>	<b>160</b>
<b>R<sub>p1.0</sub></b>	<b>N/mm<sup>2</sup></b>	<b>305</b>	<b>270</b>	<b>225</b>	<b>205</b>	<b>190</b>
<b>R<sub>m</sub></b>	<b>N/mm<sup>2</sup></b>	<b>635</b>	<b>615</b>	<b>560</b>	<b>525</b>	<b>510</b>

## Corrosion Resistance

### UNIFORM CORROSION

Uniform corrosion is characterised by a uniform attack of the steel surface in contact with the corrosive medium. This type of corrosion may occur in certain acids and hot concentrated alkaline solutions. Impurities in the form of halides, such as chloride, bromide or iodide ions, can accelerate the uniform corrosion rate. This applies especially to cases where halides occur in non-oxidising acids. Resistance to uniform corrosion is normally considered good if the corrosion rate is less than 0.1 mm/year.

In pure dilute sulphuric acid (Figure 1) 254 SMO is much more resistant than 17-12-2.5 (type 316) but is somewhat less resistant in acids of high concentration. In sulphuric acid contaminated with chloride ions (Figure 2) 254 SMO shows the highest resistance. The high nickel content of 254 SMO ensures that the corrosion rate will not increase steeply if the temperature that is indicated by the curve for 0.1 mm/year is exceeded.

Stainless steels of the type 17-12-2.5 (316) cannot be used in hydrochloric acid due to the risk of localised and uniform corrosion. However, as appears from Figure 3, 254 SMO can be used in dilute acid at moderate temperatures. Pitting need not be feared in the zone below the borderline but crevices must be avoided.

Conventional stainless steels also have a limited range of resistance in fluosilicic acid (Figure 4) and hydrofluoric acid (Figure 5). 254 SMO, on the other hand, can be used within a relatively wide range of concentrations and temperatures.

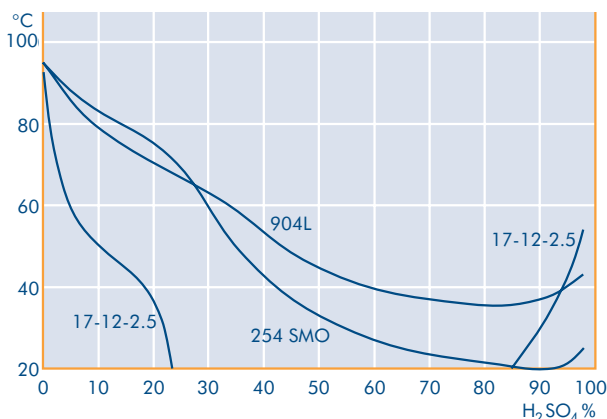


Fig. 1. Isocorrosion curves, 0.1 mm/year, in pure sulphuric acid.

## Physical Properties

Table 5

Typical values		20	100	200	300	400°C
<b>Density</b>	<b>g/cm<sup>3</sup></b>	<b>8.0</b>	–	–	–	–
<b>Modulus of elasticity</b>	<b>kN/mm<sup>2</sup></b>	<b>200</b>	<b>195</b>	<b>185</b>	<b>178</b>	<b>170</b>
<b>Thermal expansion 20–T °C</b>	<b>x10<sup>–6</sup>/°C</b>	–	<b>16</b>	<b>16</b>	<b>16.5</b>	<b>17</b>
<b>Thermal conductivity</b>	<b>W/m°C</b>	<b>13</b>	<b>14</b>	<b>15</b>	<b>17</b>	<b>18</b>
<b>Thermal capacity</b>	<b>J/kg°C</b>	<b>500</b>	<b>520</b>	<b>540</b>	<b>555</b>	<b>570</b>
<b>Electrical resistivity</b>	<b>nΩm</b>	<b>850</b>	<b>900</b>	<b>950</b>	<b>1030</b>	<b>1100</b>

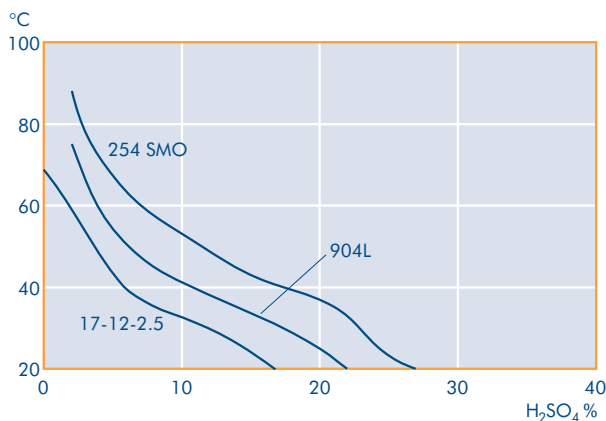


Fig. 2. Isocorrosion curves, 0.1 mm/year, in sulphuric acid containing 2000 ppm of chloride ions.

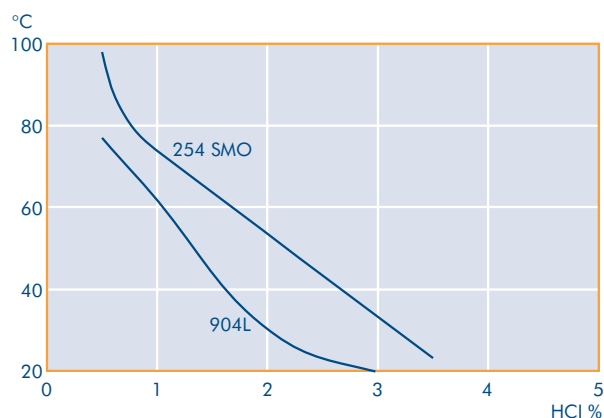


Fig. 3. Isocorrosion curves, 0.1 mm/year, in hydrochloric acid.

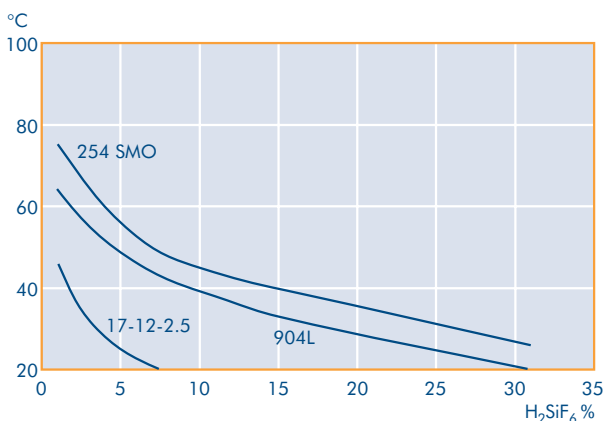


Fig. 4. Isocorrosion curves, 0.1 mm/year, in fluosilicic acid.

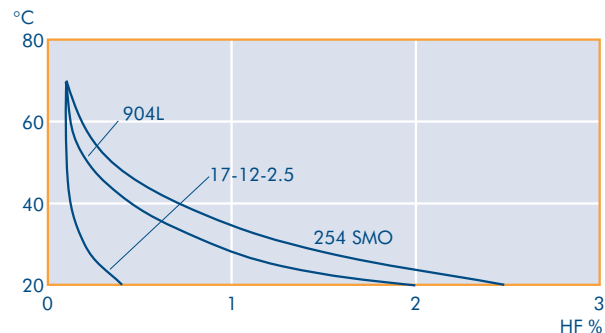


Fig. 5. Isocorrosion curves, 0.1 mm/year, in hydrofluoric acid.

Wet process phosphoric acid (WPA) always contains corrosive impurities such as chlorides and fluorides. In cases where the acid has a high chloride content, 254 SMO, as appears from Table 6, is often more resistant than conventional stainless steels.

In strongly oxidising acids, e.g. nitric acid, steels without molybdenum of the type 18-9 (304) or 18-10L (304L) are more resistant than molybdenum-containing steels. However, as appears from Table 7, 254 SMO can be the best material choice for mixtures of nitric acid and halides.

#### UNIFORM CORROSION IN WET PROCESS PHOSPHORIC ACID AT 60°C

Table 6

Avesta Sheffield	Corrosion rate, mm/year
17-12-2.5	>5
904L	1.2
<b>254 SMO</b>	<b>0.05</b>

Composition in per cent: P<sub>2</sub>O<sub>5</sub> 54; HCl 0.06; HF 1.1;  
H<sub>2</sub>SO<sub>4</sub> 4.0; Fe<sub>2</sub>O<sub>3</sub> 0.27; Al<sub>2</sub>O<sub>3</sub> 0.17; SiO<sub>2</sub> 0.10;  
CaO 0.20; MgO 0.70

#### UNIFORM CORROSION IN PICKLING ACID AT 25°C

Table 7

Avesta Sheffield	Corrosion rate, mm/year
17-12-2.5	>5
904L	0.51
<b>254 SMO</b>	<b>0.31</b>

Composition in per cent: HNO<sub>3</sub> 20; HF 4.

**More information about corrosion resistance in specific media can be found in the Avesta Sheffield Corrosion Handbook.**

PITTING CORROSION

Environments containing chlorides may cause pitting attacks on stainless steels. Grades with high content of chromium, molybdenum, and nitrogen possess a high resistance to pitting. Several methods are used to compare the pitting resistance of different grades. One common method is to determine the critical pitting temperature (CPT). This is the lowest temperature at which pitting occurs when the material has been exposed to a particular environment.

Pitting can easily occur in oxidising chloride environments. In laboratory conditions, it is possible to emulate oxidising conditions by electrochemical means, using a potentiostat to give the steel a high potential. The CPT can then be determined by varying the temperature of the test solution and observing when pitting attacks occur. Using a specially designed test cell, known as the “Avesta Cell”, it is possible to eliminate the risk of crevice corrosion altogether, so that the CPT can be determined with a high degree of accuracy. Figure 6 shows some typical levels of CPT for different steel grades, obtained in a 1M NaCl solution.

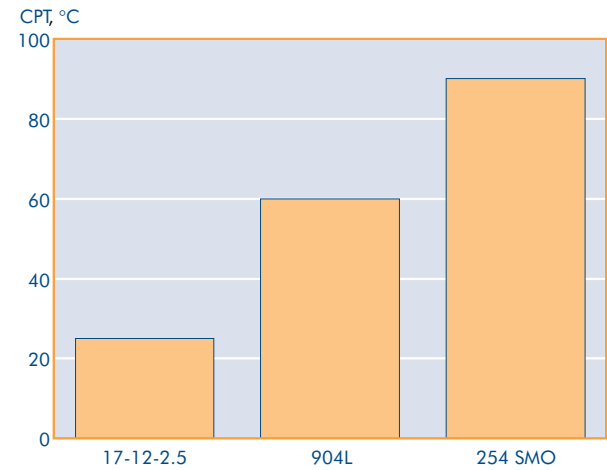


Fig. 6. Typical critical pitting temperatures (CPT) in 1M NaCl. (The Avesta Cell method, ASTM G150).

CREVICE CORROSION

The weak point of the conventional stainless steel grades is their limited resistance to crevice corrosion. As is the case with pitting, crevice corrosion occurs primarily in chloride solutions, and in this case especially under gaskets, deposits or fouling of various kinds. Figure 7 shows the resistance to crevice corrosion of different materials in a greatly accelerated test. The result is measured as the critical crevice corrosion temperature (CCT) at which crevice corrosion is initiated in this very aggressive solution. 254 SMO is superior to the other stainless steel grades.

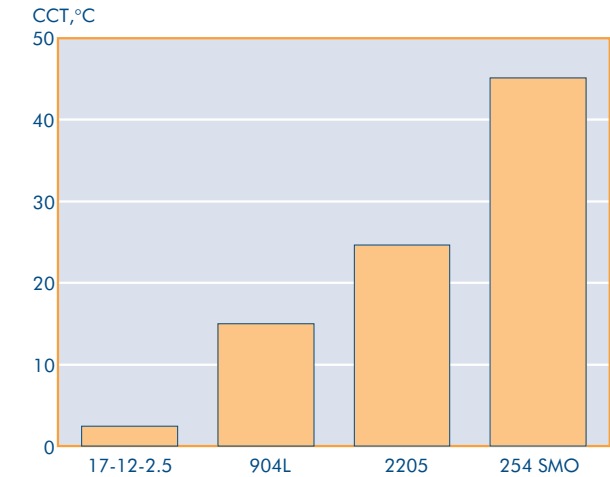


Fig. 7. Critical crevice corrosion temperatures (CCT) in 6% FeCl<sub>3</sub> (MTI-2). PTFE blocks.

CREVICE CORROSION IN SEAWATER

Natural seawater contains biological organisms, which rapidly produce a mucous layer (biofilm) on a steel surface. This film increases the corrosion potential of the steel, which in turn increases the risk of crevice corrosion. This explains why natural seawater is more corrosive than sterile chloride solutions. Natural seawater is at its most corrosive between 25–30°C, whilst at 30–35°C the biofilm becomes less active. Tables 8 and 9 show the result of two tests in natural seawater. The welded samples had strongly tightened plastic washers, covering both weld and base material.

Contrary to the conventional stainless steel grades, 254 SMO has an excellent resistance in natural seawater. In many seawater systems the water is chlorinated, mostly by use of hypochlorite, to avoid fouling. Hypochlorite and chlorine are strong oxidisers and increase the corrosion potential of the steel surface to higher levels than is normal in natural seawater, thus increasing the risk of crevice

CREVICE CORROSION  
IN FILTERED ATLANTIC WATER

Table 8

Avesta Sheffield	No. of specimens attacked	Deepest attack, mm
17-12-2.5	3 out of 3	1.1
904L	3 out of 3	1.5
254 SMO	0 out of 15	0

Flow rate < 0.1 m/s. 3 months at 25°C.

CREVICE CORROSION  
IN ATLANTIC WATER

Table 9

Avesta Sheffield	No. of specimens attacked	Deepest attack, mm
17-12-2.5	3 out of 3	>3
904L	3 out of 3	1.1
254 SMO	2 out of 12	0.09

The samples have been immersed in the sea for 18 months at ambient temperature (5–30°C).

corrosion initiation further. Chlorinated water becomes more aggressive with increasing temperature. The highest chloride content which can be used without risking crevice corrosion in flange connections depends on temperature. Practical experience indicates that 254 SMO can be used for seawater handling systems at a residual chlorine content of 1 ppm and temperatures up to 30°C. However, much higher chlorine concentrations can be accepted if intermittent chlorination is used.

### STRESS CORROSION CRACKING

Stress corrosion cracking (SCC) may occur if a material is subjected to tensile stress while in contact with a corrosive medium, usually resulting in the formation of cracks. Tensile stress may be caused by fabrication processes such as welding and bending.

Ordinary austenitic steels may be sensitive to stress corrosion cracking in a chloride rich environment at temperatures above 60°C. The resistance to SCC is enhanced by elevated levels of nickel and molybdenum in the steel.

One method to evaluate the resistance to SCC is the drop evaporation test (DET). In this test, a uniaxially stressed specimen is initially resistance heated, upon which a dilute sodium chloride solution (0.1M NaCl) is dripped at a rate of 6 drops per minute. The resistance heating is adjusted to allow each drop to evaporate completely before the next drop hits the specimen. Due to the cooling effect of the solution, the surface temperature of the specimen varies slightly around 100°C. The main purpose of the drop evaporation test is to simulate wetting/evaporation cycles often found in practical applications.

The recorded result is the time to fracture of the specimen. The threshold value of the stress which leads to fracture within 500 hours is determined and expressed as a percentage of the yield strength at 200°C for the steel grade. Figure 8 shows such test results for some stainless steels.

According to this test 254 SMO possesses very good resistance to stress corrosion cracking. However, the steel is not immune against this type of corrosion, it suffers attack in

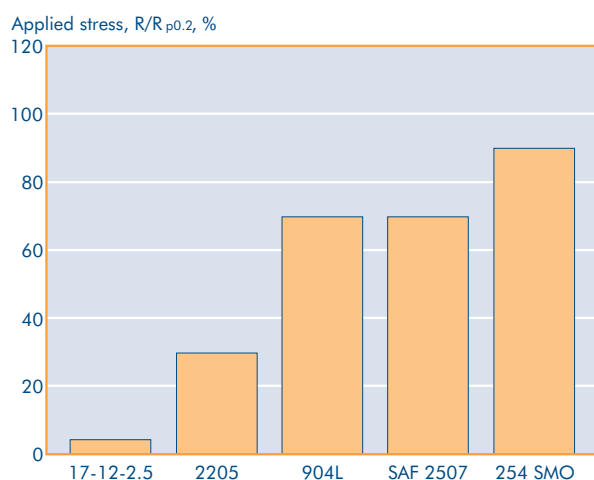


Fig. 8. SCC - threshold stresses determined using the DET method.

boiling 45% magnesium chloride which is used as the corrosive medium in the severest test procedure. Ferritic steels alloyed with nickel or copper, and ferritic-austenitic steels also fail in this test.

### SULPHIDE STRESS CORROSION CRACKING

Hydrogen sulphide may cause embrittlement in ferritic structures and sometimes also in heavily cold worked duplex and austenitic structures. When both hydrogen sulphide and chlorides are present in significant concentrations, the susceptibility to corrosion induced cracking of stainless steels is increased to some degree. Such sour environments are typically found in oil and gas production.

NACE standard MR0175-95 specifies the material requirements for resistance to sulphide stress corrosion cracking (SSCC) for equipment used in oil and gas production. In this standard 254 SMO (UNS S31254) is accepted in both annealed and cold worked condition to a maximum hardness level of 35 HRC. For common austenitic types like 304 and 316 a maximum hardness of 22 HRC is specified. Also, this standard does not allow cold working of the common types in order to increase mechanical strength.

### INTERGRANULAR CORROSION

254 SMO has a very low carbon content. This means that there is very little risk of carbide precipitation in connection with heating. The steel passes the Strauss test (ASTM A 262 practice E) even after one hour of sensitising at 600–1000°C.

However, due to the high alloy content of the steel, intermetallic phases can precipitate at the grain boundaries in the above mentioned temperature range (see under Structure). These precipitates do not involve a risk of intergranular corrosion in the corrosive media where this steel is used. Thus welding can be carried out without risk of intergranular corrosion. However, in hot concentrated nitric acid these phases can give rise to intergranular corrosion in the heat-affected zone.

### EROSION CORROSION

As distinguished from copper alloys, stainless steels have no real limitation as regards flow velocities in, for instance, seawater piping systems. For copper-nickel (e.g. CuNi 90/10), however, erosion corrosion can occur at velocities above 3 metres per second. In stagnant and flowing waters the copper alloys are attacked by sulphide contaminants from decomposition of organic and other matter. 254 SMO is not subject to these types of attack.

### GALVANIC CORROSION

254 SMO is not affected by galvanic corrosion when connected to titanium in seawater. However, copper alloys are attacked, like carbon steel, when in contact with 254 SMO (or titanium). The galvanic attack is increased when the area of 254 SMO is increased in relation to the area of the copper alloy as illustrated by Figure 9. If the seawater is chlorinated the galvanic corrosion rate is reduced.



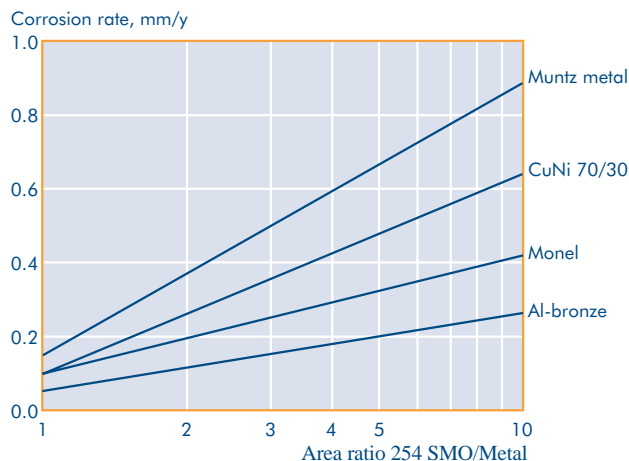


Fig. 9. Galvanic corrosion of copper alloys in slow moving seawater at ambient temperature.

## Fabrication

### HOT FORMING

Hot working should be carried out in the temperature range 1200–1000°C. Higher temperatures will reduce the workability. Fairly heavy scaling occurs at temperatures exceeding 1150°C.

To ensure dissolution of possible precipitates of secondary phases from the hot forming, the subsequent heat treatment should take place at min. 1150°C. The material should then be cooled as quickly as possible. A cooling rate that is too slow can result in reduced corrosion resistance.

### COLD FORMING

254 SMO possesses good cold formability. Bending, pressing and other forming operations which are used in fabrication can be carried out without difficulty. Practical experience obtained from the pressing of heat exchanger plates has been very favourable. The steel work-hardens rapidly. This is demonstrated in Figure 10. Spinning operations may require intermediate annealing, especially if the blanks contain welds.

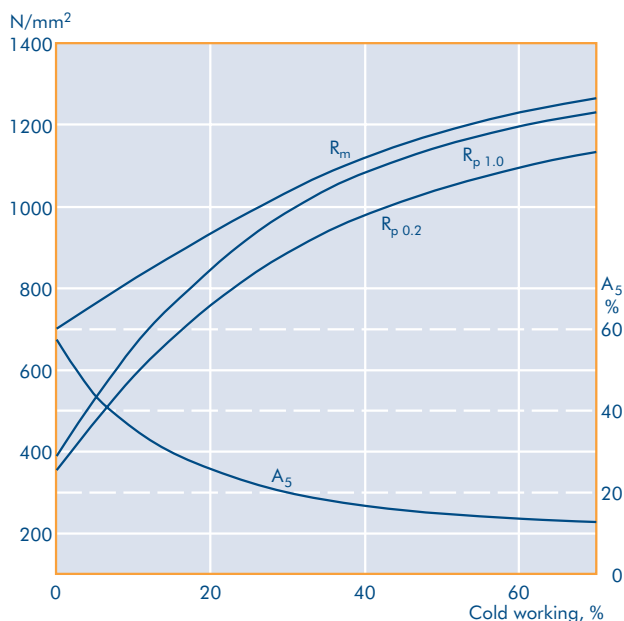


Fig. 10. 254 SMO work-hardening at cold rolling.

## MACHINING

Highly alloyed austenitic steels, such as 254 SMO, are generally more difficult to machine than conventional austenitic stainless steels such as 17-12-2.5 (type 316). 254 SMO requires higher cutting forces and causes more rapid tool wear than lower alloyed stainless steels, due to its higher strength and alloy content.

The relative machinability of 254 SMO compared to other stainless steels is illustrated by the machinability index in Figure 11. This index, which rises with increased machinability, is based on a compounded evaluation of test data from several different machining operations. It gives a rough indication of the machinability of different stainless steel grades in relation to that of 17-12-2.5 (type 316). It should be noted that the machinability index does not describe the relative difficulty of machining with cemented carbide or high speed steel tools.

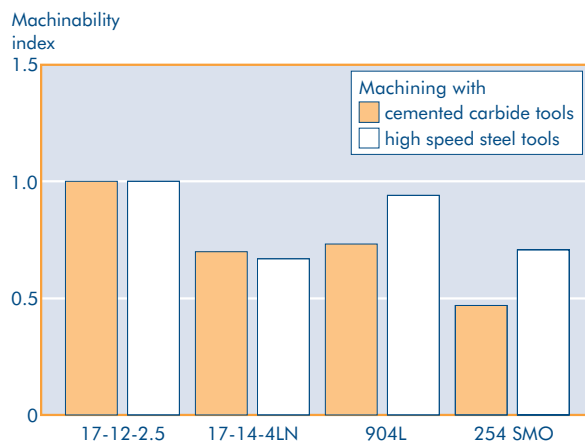


Fig. 11. Relative machinability for some stainless steel grades.

## WELDING

254 SMO possesses good weldability. When delivered, sheets and plates in 254 SMO have a homogeneous composition. Remelting of the parent metal, such as during welding without filler metal, may cause microscale variations in composition for elements such as chromium, nickel and, particularly, molybdenum. This phenomenon occurs in all highly alloyed stainless steels. These variations may reduce the pitting resistance of the weld. GTA- and plasma-arc welding without filler metal should therefore be avoided. Avesta P12 and P16 welding consumables have a very high alloy type of composition, Table 10. These filler metals will give a weld with a pitting resistance comparable to the base metal.

## WELDING CONSUMABLES

Table 10

	Weld metal, typical composition, %						
	C	Si	Mn	Cr	Ni	Mo	Nb
<b>Avesta P12</b>							
<b>Welding wire</b>	0.03	0.25	0.2	21	63	9	3
<b>Covered electrode</b>	0.02	0.5	2	21	60	9	2
<b>Avesta P16</b>							
<b>Welding wire</b>	0.02	0.2	0.5	23	60	16	–
<b>Covered electrode</b>	0.02	0.3	0.5	25	59	15	–

The following welding instructions should be observed:

1. The material may not be subjected to abrasive contact with copper/brass items. Penetration of copper/zinc into the grain boundaries can give rise to crack formation.
2. Avesta P12 or P16 welding consumables should be used for all welding methods. GTA- and plasma-arc-welding without filler wire should be avoided in cases where post-weld annealing is impossible.
3. Ignite the electrode in the joint since ignition burns beside the weld can give rise to corrosion attacks.
4. Weld with low heat input, i.e. small diameter filler metal. Weaving should be avoided in horizontal position. Do not use unnecessarily high amperages or thicker electrodes than necessary.
5. In multi-pass welding the workpiece should be allowed to cool to approx. 100°C before the next pass is welded.
6. If a welding pass is terminated too abruptly, so-called pipes or crater cracks can occur in this material just as in other austenitic steels. Such defects must be removed carefully by grinding. Pipe formation can be avoided if the electrode is moved backwards somewhat through the molten pool and lifted gently through the slag.
7. Ensure during GTA- and plasma-arc-welding that the filler wire is fed as continuously and as evenly as possible to avoid variations in composition.
8. Heat treatment is normally not necessary after welding. However, welding without filler metal should be followed by solution annealing at 1150–1200°C and subsequent rapid cooling in order to achieve the best possible corrosion resistance.
9. To ensure optimum corrosion resistance the welds should be cleaned, preferably by pickling.

Further information regarding joint designs, welding methods, etc., is found in the Avesta Welding leaflet How to weld 254 SMO, Information No. 9556.

## Products

### HOT ROLLED PLATE AND COIL

Dimensions according to Avesta Sheffield AB's manufacturing programme.

### COLD ROLLED PLATE, SHEET AND STRIP

Dimensions according to Avesta Sheffield AB's manufacturing programme.

### BAR AND FORGINGS

Delivered by Avesta Valbruna AB, Karlstad, Sweden.

### TUBE AND PIPE

Welded tube and pipe is manufactured by AST (Avesta Sandvik Tube AB), Sweden. Seamless tube and pipe is delivered by the licensees AB Sandvik Steel, Sandviken, Sweden, and Sumitomo Metal Industries Ltd, Japan.

### FITTINGS

Welding fittings, threaded fittings and flanges are manufactured by Calamo Nords AB, Molkom, and Avesta ABE AB, Örnsköldsvik, Sweden.

### WIRE ROD

Wire rod is delivered by the licensed producer Fagersta Stainless AB, Fagersta, Sweden.

### WIRE

Cold drawn wire is delivered by the licensees Fagersta Stainless AB, Fagersta, Sweden and AB Sandvik Steel, Sandviken, Sweden.

### WELDING CONSUMABLES

Covered electrodes as well as wire for automatic welding (GMAW, GTAW and SAW) of type Avesta P12 and Avesta P16 are produced by Avesta Welding AB, Avesta, Sweden.

### CASTINGS

Several foundries are licensed to produce 254 SMO castings. They are located in the countries listed below. Addresses can be obtained from your nearest Avesta Sheffield representative.

Canada	Norway
Germany	Philippines
Great Britain	Spain
Italy	Sweden
Japan	USA

## **PRODUCT STANDARDS AND APPROVALS**

EN 10088-2

Stainless steels. Flat products for general purposes

EN 10088-3

Stainless steels. Long products for general purposes

EN 10028-7

Stainless steels. Flat products for pressure purposes.

EN 10272

Stainless steel bars for pressure purposes

EN 10283

Corrosion resistant steel castings

Approval

Arbetsarskyddsstyrelsen

NGS 1605

Nordic Rules for application of the non-standard steel Avesta Sheffield 254 SMO (plate, sheet, strip, bar and forgings for pressure vessels)

Norsok M-CR-630

Material data sheets for 6Mo stainless steels

VdTÜV WB 473

Austenitischer Walz- und Schmiedestahl X2 CrNiMoCuN 20 18 6. Blech, Band, Schmiedestück, Stabstahl für Druckbehälter

TÜV Wien WB 182

Austenitischer Walz- und Schmiedestahl X2 CrNiMoCuN 20 18 6. Blech, Band, Schmiedestück, Stabstahl für Druckbehälter

ASTM A182/ASME SA-182

Forged or rolled alloy-steel pipe flanges, forged fittings, and valves and parts for high temperature service

ASTM A193; ASTM A194

Alloy and stainless steel bolts and nuts for high pressure and high temperature service

ASTM A240/ASME SA-240

Heat resisting Cr and CrNi stainless steel, plate, sheet, strip for fusion-welded unfired pressure vessels

ASTM A249/ASME SA-249

Welded austenitic steel boiler, superheater, heat exchanger and condenser tubes

ASTM A269

Seamless and welded austenitic stainless steel tubing for general service

ASTM A276

Stainless and heat resisting steel bars and shapes

ASTM A312/ASME SA-312

Seamless and welded austenitic stainless steel pipe

ASTM A351/ASME SA-351

Steel castings, austenitic, for high temperature service

ASTM A358/ASME SA-358

Electric fusion-welded austenitic CrNi alloy steel pipe for high temperature

ASME SA-403

Wrought austenitic stainless steel piping fittings

ASTM A409/ASME SA-409

Welded large diameter austenitic steel pipe for corrosive or high temperature service

ASTM A473

Stainless and heat resisting forgings

ASTM A479/ASME SA-479

Stainless and heat resisting steel wire, bars and shapes for use in boilers and other pressure vessels

ASTM A743; ASTM A744

Castings, iron-chromium, iron-chromium-nickel for general and severe service

ANSI B31.1

Power piping

ANSI B16.34\*

Valves, flanged and butt welding ends

NACE MR0175

Sulphide stress cracking resistant material for oil field equipment

\*application to be approved

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